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TITLE:

Synthesis and ir spectra of some oxygen-containing

organotin and organosilicotin compounds

PERIODICAL:

Zhurnal obshchey khimii, v. 32, no. 12, 1962,

4007-4012

TEXT: The synthesis and properties of the following are described: $(CH_3)(\underline{iso}-C_4H_9)(C_6H_5)SiOSn(\underline{iso}-C_4H_9)_3$, $[(\underline{iso}-C_4H_9)_3SnO]_2$ $Si(C_2H_5)_2$, $[(CH_3)(\underline{iso}-C_4H_9)(C_6H_5)SiO]_2$ $Sn(\underline{n}-C_4H_9)_2$. The ir spectra of these compounds and of the hexaelkylstannoxides $R_3SnOSnR_3$ where $R = \underline{iso}-C_4H_9$, $\underline{n}-C_7H_{15}$ and the dialkylstannones R_2SnO where $R = C_2H_5$, $\underline{n}-C_3H_7$, $\underline{iso}-C_5H_{11}$, $\underline{n}-C_7H_{15}$ are given in order to study the spectroscopic characteristics of the \rightarrow SnO_3 , \rightarrow SiO_4 , and \rightarrow $SnOS_4$ \leftarrow bonds. The organosilicotin compounds were synthesized, respectively, by the reaction of the appropriate hexaelkylstannoxide

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Synthesis and ir spectra

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with the appropriate silanol and silanediol and of the appropriate dialkylstannone with the appropriate silanol. By comparing the ir spectra of all these compounds and those of stannic oxide and tetra-iso-butylstannane, assignments of the main bands are made. The absorption frequency of the Sn—0 bond in the Sn—0—Sn group is about 780 cm⁻¹. In the Si—0—Sn group, the Sn—0 frequency is lowered to 720 cm⁻¹ and that of Si—0 ro 980 cm⁻¹. In the case of the dialkylstannones, strong bands at 570 ± 5 and 415 ± 10 cm⁻¹ are assigned to the Sn—0—Sn group, confirming their polymeric nature, [R₂SnO]_X. There are 3 figures and 1 table.

ASSOCIATION:

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